

26/11/24

Time: 2 ½ Hours.

Total Marks: 75

N.B.:

1. All questions are compulsory.
2. Figures to the right indicate full marks.
3. Draw neat diagrams wherever necessary.
4. Symbols have usual meaning unless otherwise stated.
5. Use of non-programmable calculator is allowed.

Constants: Boltzmann Constant: $k = 1.38 \times 10^{-23} \text{ J}^\circ\text{K}$
Planck's Constants: $h = 6.63 \times 10^{-34} \text{ J-sec}$

1. (A) Attempt any one: -
 - (i) State and explain probability theorems. 10
 - (ii) Explain the Normal or Gaussian distribution. 10
- (B) Attempt any one: -
 - (i) Write a short note on "Sample Space" 5
 - (ii) Three coins are tossed; what is the probability that two are tails and one head? What is the probability of getting two consecutive tails and then a head? If there was at least one tail, what is the probability of all tails? 5
2. (A) Attempt any one: -
 - (i) Explain terminology and notation of a complex number. Find x, y, r, θ of given complex number z . Plot the number and label it and find its complex conjugate. 10
 - a) $z = 3 + 3i$
 - b) $z = 2 \left(\cos \frac{\pi}{6} + i \sin \frac{\pi}{6} \right)$
 - (ii) Explain hyperbolic functions of complex numbers. Using these prove that 10
 - a) $\cosh^2 z - \sinh^2 z = 1$
 - b) $\frac{d}{dz} \cosh z = \sinh z$
- (B) Attempt any one: -
 - (i) Prove: $\cos z = 2$ 5
 - (ii) A particle moves in (x, y) plane so that its position (x, y) as a function of time t is given by 5

$$z = 5e^{i\omega t}$$
 Calculate the magnitude of velocity and acceleration.
3. (A) Attempt any one: -
 - (i) Define partition function. Obtain an expression for translational partition function. Find translational partition function for A_r (mass $6.63 \times 10^{-26} \text{ kg}$) confine to a volume of 1 liter at 298 K. 10
 - (ii) Derive an expression for total energy for two-level system. Determine the total energy of an ensemble consisting of N particles that have only two energy levels separated by energy $h\nu$. 10

- (B) Attempt any one: -
- (i) Write a short note on degeneracy of energy states. 5
 - (ii) What is the weight associated with the configuration corresponding to observing 40 heads after flipping a coin 100 times? How does this weight compare to that of the most probable outcome? 5
4. (A) Attempt any one: -
- (i) Imagine N identical but distinguishable balls are randomly distributed within a large box that is divided into k cells, each with a different area. How can we demonstrate that the number of balls in each cell will be proportional to the cell's area? 10
 - (ii) How can we derive the Bose-Einstein distribution law that describes how bosons occupy various energy states in a thermodynamic system? 10
- (B) Attempt any one: -
- (i) Derive the expressions for the mean velocity of gas molecules that adhere to the Maxwell-Boltzmann distribution law. 5
 - (ii) Three identical particles can be in any of the five states. What are the number of possible ways of distributing them in various states according to Maxwell-Boltzmann (MB), Bose-Einstein (BE), and Fermi-Dirac (FD) statistics? 5
5. Attempt any Five: -
- (i) There are 10 chairs in a row and 8 people to be seated. In how many ways can this be done? 3
 - (ii) Define probability of an event. A three-digit number is selected "at random". What is the probability that all three digits are same? 3
 - (iii) Find the absolute value of given complex number z . 3
- $$z = \frac{25}{3+4i}$$
- (iv) Evaluate: $i^{\ln i}$ 3
 - (v) What is the difference in energy between $n = 2$ and $n = 1$ states for molecular oxygen (mass $= 5.31 \times 10^{-26}$ kg) constrained by a one-dimensional box having length of 1.00 cm? 3
 - (vi) Using the given partition function $q = 1.58$, calculate the probability of an oscillator occupying the first three energy levels ($n = 0, 1 \& 2$) 3
 - (vii) A large box of area 200 m^2 is divided into small square cells. If 10^6 balls are thrown at random in the box, find the most probable number of balls which fall in the square cell of side 10 cm. 3
 - (viii) A black body cavity is cubical with each side of length 10 cm. calculate the number of modes in the wavelength region 6000 \AA to 6002 \AA . 3