

[Time: 3 Hours]

[Marks: 100]

- All questions are compulsory.
- All questions carry equal marks.
- Figures to the right indicate full marks.
- Use of log table/ non-programmable calculator is allowed.

- Q.1 Attempt any four of the following.** 20
- A) Give the full form of NGP. Explain with a suitable example the effect of NGP on kinetics and stereochemistry of the reaction. 5
- B) a) Distinguish between acidity and electrophilicity. 3  
 b) What is pyrolytic elimination of acetates? 2
- C) What are pericyclic reactions? Name different types of pericyclic reactions. Explain any one of it with a suitable example. 5
- D) a) Explain acyl nucleophilic substitution reaction with suitable example. 3  
 b) Complete the following reaction and name the reaction involved. 2  

$$\text{CH}_3\text{COOCH}_3 + \text{NaOH} \longrightarrow ?$$
- E) What is photoreduction? Explain the mechanism of photoreduction of benzophenone. 5
- F) a) Explain the Norrish type-II reaction of 2-pentanone. 3  
 b) Write a note on photosensitization reaction. 2
- Q.2 Attempt any four of the following.** 20
- A) Write a note on stereochemistry of allenes. 5
- B) Define molecular chirality. State whether the following compounds are optically active or optically inactive. 5
- i)

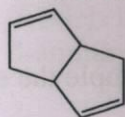
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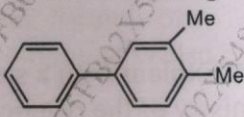
iv)
- C) Give the Skraup synthesis of quinoline. Write the reaction of quinoline with n-butyl lithium. 5
- D) What is the action of following reagents on pyridine N-oxide? 5
- $\text{Br}_2$ , Oleum,  $70^\circ\text{C}$
  - $\text{SO}_2\text{Cl}_2$
  - $\text{NaNH}_2$  in Liq.  $\text{NH}_3$
  - Conc.  $\text{HNO}_3 + \text{Conc. H}_2\text{SO}_4$ ,  $160^\circ\text{C}$
  - n-BuLi
- E) a) Write a note on biopesticides. 3  
 b) Draw the resonating structures of Pyridine-N-oxide. 2
- F) What are agrochemicals? Mention its two advantages. Write the synthesis of Indole-3-acetic acid. 5

**Q.3 Attempt any four of the following.** **20**

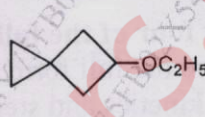
- A) Illustrate any two principles of Green Chemistry with suitable example. **5**  
 B) Draw the structure of following organic compounds; **5**  
 i) Bicyclo[2.2.2]octan-2-one      ii) Hepta-2,3,4-trienoic acid  
 iii) 4,4'-Dinitrobiphenyl      iv) 3,4-Difluoroquinoline  
 v) 1-Ethyl-isoquinoline
- C) Write IUPAC name of the following compounds; **5**



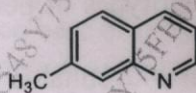
i)



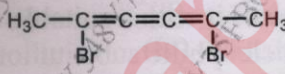
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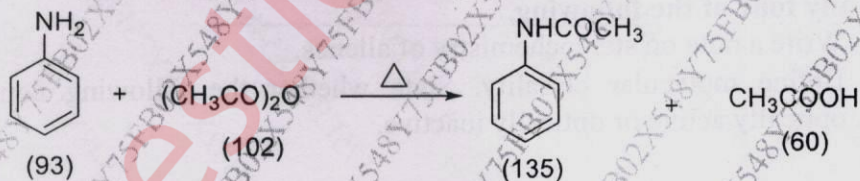


iv)



v)

- D) Explain the following terms with suitable example **5**  
 i) Green reagent  
 ii) Green catalyst
- E) Calculate the % atom economy and theoretical yield of the following reaction **5**  
 if the given weight of aniline is 0.5g.



- F) Write preparation of 1-butanol and 2-butanol using a suitable Grignard reagent. **5**

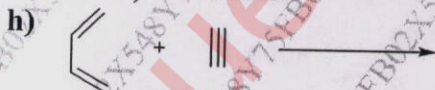
**Q.4 Attempt any four of the following.** **20**

- A) Write the synthesis of citral starting from methyl heptenone. **5**  
 B) Explain chromophore-auxochrome interaction in UV-visible spectroscopy with suitable examples. **5**  
 C) What is Molecular ion peak? Give the mass fragmentation of acetone and 2-methylbut-2-ene. **5**  
 D) Write the synthesis of Adrenaline from 2,3-Dihydroxy benzaldehyde. **5**  
 E) What are Isoprene and special isoprene rules? Write the harmful effects of Nicotine. **5**  
 F) Explain the use of Hoffmann's exhaustive methylation and degradation in finding structure of alkaloids. **5**

Q.5 A) Select the correct option and complete the following statements: (any five)

5

- a) Cope elimination is observed in.....
  - i) N-substituted amide
  - ii) Aromatic ketoxime
  - iii) Tertiary amine oxide
- b) All nucleophiles are.....
  - i) Lewis acid
  - ii) Lewis base
  - iii) Neutral
- c) 1,3,5-Hexatriene undergoes electrocyclic reaction to for.....
  - i) 1,3-Cyclohexadiene
  - ii) Cyclohexane
  - iii) 1,4-Cyclohexadiene
- d) Basicity is a .....term.
  - i) Kinetic
  - ii) Thermodynamic
  - iii) Photochemical
- e) Phosphorescence involves the transition of.....
  - i) Singlet-Singlet
  - ii) Triplet-Triplet
  - iii) Triplet-Singlet
- f) Conversion of Allyl benzene to phenyl cyclopropane is an example of ..... reaction.
  - i) di- $\pi$  Methane
  - ii) Norrish type-I
  - iii) Norrish type-II
- g) A triplet state has a .....energy than the singlet state.
  - i) Low
  - ii) High
  - iii) Same



The above reaction is.....

- i)  $(2\pi + 2\pi)$  Cycloaddition
- ii)  $(4\pi + 2\pi)$  Cycloaddition
- iii) Group Transfer reaction

Q.5 B) State whether true or false: (any five)

5

- a) Meso-tartaric acid is optically active.
- b) Cummulenes with odd number of double bonds show optical isomerism.
- c) Atropisomerism is due to restricted rotation around C-C single bond in biphenyls.
- d) DDT and BHC belong to the class of insecticides.
- e) Karanja oil is not a biopesticide.
- f) Bromination of pyridine-N-oxide takes place at 4-position
- g) Isoquinoline is also known as 2-azanaphthalene.

**Q.5 C) Fill in the blanks with correct alternatives given in the bracket: (any five)** **5**  
 (Biphenyl, Mannich,  $\text{KMnO}_4$ , Methyl Chloride, three components, Nine, Vinylic, linear,  $\text{NaSH}$ , Acid, Twelve, Benzylic)

- a) Methyl Lithium on treatment with .....forms ethane.
- b) The reduction of m-Dinitrobenzene by using.....gives m-nitroaniline.
- c) o-Chlorotoluene on treatment with .....gives o-Chlorobenzoic acid.
- d) Green Chemistry is based on.....principles.
- e) Biginelli reaction is .....type of reaction.
- f) NBS is .....brominating agent.
- g)  $(\text{C}_6\text{H}_5)_2$  is a molecular formula of .....
- h)  $\beta$ -amino carbonyl compound is also called..... base.

**Q.5 D) Match the columns: (any five)** **5**

**Column A**

- a) Turpentine oil
- b) Leuco base of crystal violet
- c) Oxidation of Citral
- d) Chromophore
- e) Eicosanoids
- f) Pinner's work
- g) Mass spectrum

**Column B**

- (i) Colour bearing group
- (ii) m/e value
- (iii) Structure of Nicotine
- (iv)  $\alpha$ -Pinene
- (v) Colourless
- (vi) Geranic acid
- (vii) Prostaglandins
- (viii)  $-\text{OH}$
- (ix) Lactic acid