Paper / Subject Code: 40322 / Industrial & Engineering Chemistry II

1T00534 - S.E.(Chemical Engineering)(SEM-IV)(Choice Base Credit Grading System) (R- 20-21) (C Scheme) / 40322 - Industrial &

Engineering Chemistry II

QP CODE: 10014220 DATE: 12/12/2022 **Time: 3 hours**Max. Marks: 80

Time. 5 hours

N.B: 1

- 1. Question.No.1iscompulsory.
- 2. Attempt any three questions from Q.No.2 to Q.No.6

Q1 Solve any Four out of Six

(20 Marks) (5 Marks each)

- A Define catalyst. What is the criteria for selecting material as a catalyst?
- B Explain Electrophoresis in detail & how is it related to zeta potential.
- C What are advantages of TLC over other chromatography techniques?
- D What do you understand by Differential thermal Analysis (DTA)? Give ANY 3 applications.
- E Give an account of Reformatasky reaction.
- F Explain dielectric constant of ionising solvents. Explain importance of non -aqueous solvents.

 $\mathbf{Q2}$

(20 Marks) (5 Marks each)

- A Explain Dorn effect in case of colloids.
- B What are non-aqueous solvents? Give the levelling effect of solvents.
- C Explain Chemical shift & Shielding & deshielding of protons in NMR spectroscopy.
- D Explain keto enol tautomerism in case of acetoacetic ester with any one preparation method.

Q3

(20 Marks) (5 marks each)

- A What is Rf value? Explain it with concept of paper chromatography.
- B Give applications of IR spectroscopy in detail.
- C Describe acid base catalysis in detail.
- D Explain conversion of ketooximes to amide.

Q4

(20 Marks) (5 marks each)

- A What is meant by thermogravimetry? Give any two applications in detail.
- B Explain (Benzil \rightarrow KOH \rightarrow -----) in detail, with reaction mechanism.
- C What do you mean by Surfactants? Explain how are they useful in detergents.
- D Give an account of HPLC with 2 applications.

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Q5 (20 Marks) (5 Marks each)

- A Explain Complex reaction & precipitation reaction in Liq ammonia.
- B Explain application of colloids in pesticides.
- C Describe Gas chromatography (Principle & 2Applications).
- D Explain Intermediate compound theory for catalysts.

Q6 (20 Marks) (5 Marks each)

- A Explain origin of charge on colloids.
- B Give any 3 applications of UV spectroscopy.
- C What is importance of non aqueous solvents? Give Acid-base & Redox reactions in Liq SO2.
- D Give preparation & properties of Malonic ester.

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