

14/11/2024 CHEMICAL SEM-III C SCHEME IEC-I QP CODE: 10065378

Time: 3 Hours

Marks:80

- N.B.: 1. Questions no. 1 is compulsory.
2. Attempt any three questions from remaining five questions
3. Figures to the right indicates full marks.

Q1. Attempt any four questions of the following.

[20]

- Why conductometric titrations superior over volumetric titrations? Give at least one limitation of conductometric titration.
- Give preparation of $\text{Fe}(\text{CO})_5$ molecule. Express their bonding, hybridization and Structure.
- Justify concept of chirality with necessary conditions for a molecule to be chiral.
- Distinguish between thermodynamically and kinetically controlled reactions.
- Give chemical composition of Ammonium sulphate. Illustrate any four qualities of ideal fertilizer.
- A 200 cm^3 of organic solution containing 0.35 g of certain solute. The solute is extracted three times with 20 cm^3 of ether. Find amount of solute remaining in aqueous solution and % extraction. If distribution ratio is 16.

Q2. a. Discuss effect of dilution & temperature on equivalent conductance.

[5]

b. Summarize isomerism in CN-4 coordination compounds with suitable example.

[5]

c. Give synthesis of Congo red dye with reaction. Give two applications.

[5]

d. State Nernst distribution law and explain batch extraction process in brief.

[5]

Q3. a. Discuss Michael addition reaction with mechanism.

[5]

b. Write a note on Carbonic anhydrase. Give any two functions.

[5]

c. Give synthesis of Methyl orange dye with reaction. Write two properties & applications.

[5]

d. A solute is extracted from its aqueous solution on treatment with carbon tetrachloride. A 100 cm^3 of aqueous solution containing 0.5 g of solute extracted with 20 cm^3 of carbon tetrachloride. Calculate (i) the amount of solute remaining unextracted (ii) the percentage extraction. (Given $D=410$)

[5]

- Q4.** a. Discuss carbene & its types. Comment on structure and stability. [5]
b. Why Anti conformation of n-butane is more stable than Gauche & Eclipsed? Justify with help of Newman representation formula. [5]
c. Give synthesis of Alizarin dye with reaction. Write two properties & applications. [5]
d. Give nomenclature for $\text{Na}_2[\text{Fe}(\text{CN})_4(\text{H}_2\text{O})_2]$ and $[\text{Ni}(\text{CO})_4]$. [5]
- Q5.** a. Calculate CFSE for d^4 and d^6 system in the strong field and weak field of octahedral complexes. [5]
b. Explain oxygen transportation reaction in haemoglobin. [5]
c. Discuss both Norrish type-I & Norrish type-II reactions with an example. [5]
d. Describe lanthanide separation by ion exchange method. [5]
- Q6.** a. Define Transport number. Explain w.r.t Moving boundary method. [5]
b. Explain pinacol pinacolone reaction with mechanism and one application. [5]
c. Name three main macro nutrient types of fertilizers. Discuss role of primary nutrients (N,P,K) and deficiency symptoms caused due to these. [5]
d. What is EAN? Calculate EAN for
(i) $[\text{V}(\text{CO})_6]$ (ii) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$. [5]
