## 3 Hours

	3 Hours Total Marks: 100
1 .	Attempt all questions.
, $\epsilon$	All questions carry equal marks.
2	
). 1	Draw neat labeled diagrams wherever necessary.
1.	Use of log tables and non-programmable calculator is allowed.
Š	Select the correct alternative (Any six)
<b>Y</b>	Which of the following does the virus lack?
	a. Nucleic acid b. Ribosome c. Glycoprotein d. Phospholipid
	Baltimore's classification of viruses has groups.
1	a. 5 b. 6 c. 7 d. 8
Š	The replication of hepatitis B includes which of the following stages?
)	a. Movement of the intact virus to the cellular cytoplasm for replication.
	b. Conversion of relaxed circular viral DNA into covalently closed circular (CCC)
	DNA in the nucleus.
V	c. Virions produced in the cytoplasm by cellular DNA polymerase.
7,	d. Oncogenic activity to transform liver cells.
	Which of the following is the example of a negative single-stranded RNA animal
	virus?
	a. Tobacco Mosaic Virus b. Influenza virus
3	c. Herpes Simplex Virus d. Pox Virus
	Plaque is formed by
	a. animal virus b. plant virus c. Bacteriophage
	d. both bacteriophage and plant virus
<u></u>	A positive strand of RNA virus
	a. should be converted to m-RNA before it can be translated.
	b. is not recognized by host ribosomes.
	c. should convert to double-stranded DNA and then be expressed.
	d. can be directly translated to produce viral protein.
S	An infection that results in cell death is referred to as
У.	a. exocytosis b. cytocidal effect
	c. cytopathic effect  d. bacteriostatic effect
	Nucleocapsid of a virus is made up of
N.	a. Nucleus and capsomers  b. Envelope and protein
3	c. Protein and Nucleic acid deviral nucleus
	Icosahedron symmetry of the capsid has
	a. 20 sides and 12 corners  b. 12 sides and 6 corners
	c. 8 sides and 4 corners  d. 10 sides and 5 corners
3	
	Answer the following questions: (Any Two)
	Describe the three main techniques for cultivating viruses.
	Liverstorn and their vices by the hand animal remarks and malacard by a back call
	Explain any two ways by which animal viruses are released by a host cell.  How are prions different from viruses, and what are viroids?

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	Q.2 a.	Select the correct alternative (Any six)	06
Å	2 <sup>7</sup> 1.	Which of these is a Beta-lactamase inhibitor?	
2	)'	a. Chloramphenicol b. Clavulanic acid c. Ampicillin d. Tetracycline	
29)V	2.	Which of the following classes of antibiotics bind to the 30S subunit of rRNA?	
J	2	a. Cephalosporins b. Aminoglycosides c. Rifampin d. Sulphonamides	
	3	Which of the following is a drug that resembles deoxy-GTP and inhibits the virus	6
	Ş. J.	DNA polymerase?	
8	Y S	a. Amantadine b. Acyclovir c. Rimantadine d. Zidovudine	/
VY.	10		
9	34.	Quinolones are responsible for	
	R	a. cell membrane disruption b. inhibition of cell wall synthesis	D
۵	6,	c. inhibition of DNA synthesis d. inhibition of protein synthesis	3
15	5. 💸	is the agent when added to a colony, inhibits its growth and on removal	
\(\frac{1}{2}\)	(2)	the colony regrows.	
7,		a. Bacteriostatic b. Bactericidal c. Antibiotic d. Antiseptic	
	6.	Nystatin is a drug used for the treatment of diseases caused by	4
ć	5	a. bacteria b. viruses c. fungi d. protozoa	1
	7.	Which antimicrobial agent binds to DNA-dependent RNA polymerase and blocks	
£) 1	200	the synthesis of mRNA?	
	100	a. Rifampicin b. Penicillin c. vancomycin d. streptomycin	
	ج <sup>2</sup> 8.	Which of the following is used as a reverse transcriptase inhibitor in the treatment	5
2	\$7 and	of HIV?	7
VCA	5	a. Acyclovir b. Rimantadine c. Adenine-arabinoside d. Azidothymidine	
SV.	9	All of the following are true regarding metronidazole EXCEPT	
<i>)</i> ′		a. It inhibits alcohol dehydrogenase b. It is used to treat giardia	ć
,	(B)	c. It causes a metallic taste in the mouth d. It is useful against trichomonoas	2
E.	) م	c. It causes a metanic taste in the mount	>)
	Q.2 b.	Give an account on the following questions: (Any Two)	14
Ž,	1	The action of metabolic pathway inhibitors with a suitable example.	17
	2.	Mechanism of drug resistance in bacteria.	
_	3.	Mode of action of Penicillin and Cephalosporins on bacteria.	75
56	, J.	Wode of action of Tenerinii and Cephaiosporms on vactoria.	
10	0.2		0.0
<i>y</i>	Q.3 a.	Select the correct alternative (Any six)	06
	1.	Due to rotation and translation, macromolecules move into and out of a very small	5
20	N. C.	region in the solution this is called	3
.05		a. Brownian motion b. Inertia c. Forward motion d. Backward motion	
3			
\	₹ <sup>0</sup> 2.	Applications of dynamic light scattering include	
	100 m	a. Determination of diffusion coefficients c. Both a and b	3
8		b. Assessment of protein aggregation d. Neither a nor b	
20	300	The atoms in atomic spectroscopy are vaporized by	
$\mathcal{O}_{\lambda}$	3.7	a. Flame b. Electric energy c. Inductively coupled plasma (ICP)	
,	3	d. All of these	
		d. All of these	
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,	\$	64B44B1DFCE927CBAC3373F93F2AE45E	
		$\lambda$ ,	

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	<u></u> 4.	In Atomic emission spectroscopy, small part of the sample is vaporized so that	6
7	V.		3
700		a. Every atomic component emits light at wavelengths corresponding to changing	7
20,	20 T	energy levels	
59		b. Every atomic component emits light at no wavelength	
	(B)	c. Every atomic component does not emit light	S
E SE		d. Neither of these	45)
	کہ ہے		
MA	5.0	The primary filter in spectrofluorimetry is placed in between	
\$	37	a. Source and cell b. Cell and detector	h
	27	c. Source and detector d. Anywhere	R
6	6.	The detection of non-fluorescent compounds can be achieved by coupling a	3
(N)	R	fluorescent probe, this is called	
(A)		a. Extrinsic fluorescence  b. Intrinsic fluorescence	
. · ·	W.	c. Auto fluorescence  d. Semi fluorescence	
Α.		d. Seim muorescence	SY
3	7.	The commonly used enzyme in bioluminescence is	1,1
13	EV.	a. Amylase b. Luciferase c. Protease d. Hexokinase	
گى. س	20,		
,	√8.	The major disadvantage of spectrofluorimetry is	~ (S)
5		a. Time required b. Quenching c. Use of two monochromators	\$
Sol	, C	d. Inaccuracy	)
20	9	When the incident light beam hits a molecule in its ground state, there is a low	
2		probability that the molecule is excited and occupies the	_
	\$\frac{1}{2}	a. Next higher vibrational state  b. Next lower vibrational state	70
	· · ·	c. Heat energy d. Neither of these	3
, Or	29	d. Nother of these	
30	- E		
ž	Q.3 b.	Discuss the following: (Any Two)	14
7	1.	a. Elastic light scattering	27
	. 5	b. How molecular mass can be determined by multi-angle light scattering?	X
	2.	Explain the principle and applications of fluorescence spectroscopy.	
2	3.	Principle and applications of luminometry.	
7	K		
N. T.	Q.4 a.	Select the correct alternative: (Any six)	06
2	1.	A material that fluoresces when struck by a charged particle or high energy photon	7
(E)	C. V.	is known as	
	3	a. Scintillatorb. Inert gas c. Halogen d. Chemiluminescence	2
_	° 2.	Decay by negatron emission: Neutron>+ Negatron	20)
~\bar{\bar{\bar{\bar{\bar{\bar{\bar{	· ·	a. Electron b. Positron c. Proton d. Helium	, D
	3.	In process, changes occur in the atomic nucleus, and particles and/or	
(CV)	6	electromagnetic radiation are emitted.	
	ST	a. fluorescence b. radioactive decay c. transformation d. scattering	
<b>S</b>			
(A)	Ć		
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	4.	In chromatography, separation relies on the attraction between
	6	oppositely charged stationary phase.
2		a. size exclusion b. gel filtrationc. ion exchanger d. affinity
	5.	can be used as a matrix in ion exchange chromatography.
(A)	VCD,	a. inert gas b. water c. mineral solution d. cellulose
	6.	Relative molecular mass determination can be done by chromatography.
_		a. Affinity b. Ion exchange c. Size exclusion d. Paper
Ñ	<sup>\$</sup> 7. <u>\$</u>	exploits the unique property of extremely specific biological
B	E.C.	interactions to achieve separation and purification.
25		a. Affinity chromatography c. Size exclusion chromatography
ζ'	N. C.	b. Centrifugation d. Ion exchange chromatography
Ž	8.	In Affinity chromatography, select a ligand that
	2	a. displays poor specificity c. displays group selectivity
W.	5.6X	b. poor stability  d. do not displays group selectivity
Y. T.	9.	In affinity chromatography, purified compound is recovered from the ligand by
	45	
S	, V	a. by dilution c. by reloading sample
\$20,	S. V.	b. by water wash  d. by either specific or non-specific elution
5	3	
?	Q.4 b.	Describe the following questions: (Any Two)
	ار ای 1.	Principle, working, and applications of size exclusion chromatography
	2.	Steps in HPLC method development and add a note on any three parameters for
100		method validation
D.,	35	GM counters and their applications in the detection of radioactivity.
/		
É	Q.5	Write Short notes on the following: (Any four)
A. C.	a. 💉	Virus purification and assays
8	b.	Mode of action of macrolides.
×′	C.	Selective toxicity.
	d.	Principle of Raman spectroscopy.
8	e. <	Atomic fluorescence spectrophotometry.
20	f.	Autoradiography. Autoradiography.
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